

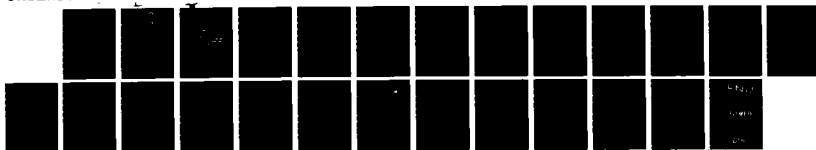
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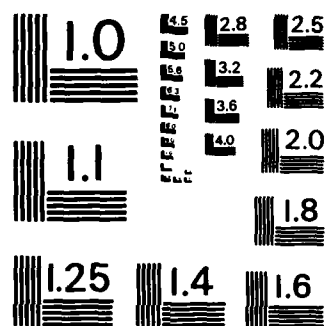
AN INTERPRETATION OF THE O2 AUGER ELECTRON SPECTRUM(U) 1/1
GEORGE WASHINGTON UNIV WASHINGTON D C DEPT OF CHEMISTRY
H SAMBE ET AL. OCT 85 TR-20 N00014-80-K-0052

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AN INTERPRETATION OF THE O₂ AUGER ELECTRON SPECTRUM

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Hideo Sambe and David E. Ramaker

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abstract

The upper electron spectrum of U_2 is interpreted by comparison with other spectra, such as the photoelectron spectrum, the electron impact mass spectrum, and the double charge transfer spectrum. Each of these four spectra obeys their own selection rules; the difference in the selection rules plays a key role in our interpretation. Under delays following a core shake-up excitation are identified for the first time. Many of the previous assignments are revised.

THE INTERPRETATION OF THE U_2 UPPER ELECTRON SPECTRUM •

Mitko Simbe and David L. Hamaker

Chemistry Department, George Washington University
Washington, D. C. 20551, USA

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1. Introduction

First Siegbahn et al. [1] and later Moddeman et al. [2] measured the Auger Electron Spectrum (AES) of U_2 with electron impact. The two spectra agree well in detail.

Figure 1 is the U_2 AES measured by Siegbahn et al.. The K1 band, which is drawn with a broken line, is taken from Moddeman's spectrum. In this paper, we assign the Auger transitions, their initial and final states, to the labelled Auger lines of Figure 1.

There are a few calculations on the U_2 AES. Hurley [3] calculated the $U_{2^{++}}$ state energies by scaling the one-electronic system U_2 . His method yielded accurate energies for the $U_{2^{++}}$, $N_{2^{++}}$, and $N_{2^{++}}$ states [3]. Keebe et al. [4] calculated the $U_{2^{++}}$ potential curves by the minimal basis full valence CI method. Dunlap et al. [5] calculated the $U_{2^{++}}$ state energies, the Auger transition probabilities, and the Auger line widths using the λ_e -method. Recently, Gregoritch and Hayes [6] calculated the $U_{2^{++}}$ state energies and the Auger transition probabilities also using the λ_e -method. These λ_e calculations cannot distinguish the various multiplet states arising from an electronic configuration.

Previous assignments of the U_2 AES were based on the total electron method above. Siegbahn et al. [1] and Moddeman et al. [2] based their assignments on the $U_{2^{++}}$

state energies calculated by Hurley [3]. Auger transition probability calculations were not available to them. Dunlap et al. [5] based their assignments on their own calculations, and so did Gregoritch and Hayes [6]. Assignments based on the λ_e calculations have serious problems, since the individual multiplet states are not distinguished. These early assignments disagree with each other more often than they agree.

The present assignments of the U_2 AES are based on the various spectroscopic data. The Fe-W Auger processes, which are defined in the following paragraph, are identified with the Photoelectron Spectrum (PES). The Fe-W Auger processes are identified with the Electron Impact Mass Spectrum (EIMS) and with the Double Charge Transfer Spectrum (DCIS). Differences in the selection rules for the three spectra give clues to the assignments. Finally, the Fe-W Auger processes are identified by using the core-level FLS.

To classify Auger processes in general, we adopt the notations by Moddeman et al. [2]. They are defined as:

Fe-W: Resonant core electron excitation into a one-hole, one-electron ($1e$) state, followed by an Auger decay into a one-hole ($1h$) state.

Fe-WW: Resonant core electron excitation into a one-hole, one-electron ($1e$) state, followed by an Auger decay into a two-hole, one-electron ($2h1e$) state.

Fe-WW: Core electron ionization into a one-hole ($1h$)

state, followed by an Auger decay into a two-hole (WW) state.

1) W-WW: (core electron ionization into a two-hole, one-electron (WW) state, followed by an Auger decay into a two-hole (WW) state.

The capital letters l and W stand for a core and a valence hole, and the small letter e stands for valence electron.

To classify electronic configurations in general, we adopt the conventional notation, that is, mh-ne for m-hole, n-electron. For instance, we write the 3h-2e configuration for the 3-hole, 2-electron configuration; 2h-1e states for 2-hole, 1-electron states; and the 2h state for the 2-hole state.

2. R1 - R7 lines

Lines R1, R2, ..., R7 in Fig. 1 behave differently from the rest. These lines are present in the electron-impact (E.E.=5keV) AES as shown in Fig. 1, but they are absent in the photon-impact (hv=1487eV) AES [2]. From this, we can conclude that the R lines originate from a neutral, discrete, core-excited state [2], that is, from Fe-W or Fe-WW processes.

High-energy (~2keV) electron-impact excitation probabilities to the neutral discrete states have been measured by FELS (electron energy loss spectra) [1,8]. These spectra show that the excitation to the $10_{\text{g}}^{-1}1\text{x}_{\text{g}}(^3\text{P}_{\text{g}})$

state at 530.8eV above the ground state predominates. We expect therefore that the R lines originate from this $10_{\text{g}}^{-1}1\text{x}_{\text{g}}(^3\text{P}_{\text{g}})$ state.

The final states of the $10_{\text{g}}^{-1}1\text{x}_{\text{g}}\text{-AES}$ and the PES are identical; in other words, the final-state binding energies of these two spectra should agree with each other. Figure 2 compares the final-state binding energies obtained from the $10_{\text{g}}^{-1}1\text{x}_{\text{g}}\text{-AES}$ [1], the $1\text{x}_{\text{g}}\text{-PES}$ [9], and the 10_{g}-PES [10]. The $10_{\text{g}}^{-1}1\text{x}_{\text{g}}\text{-AES}$ is the R line portion of the AES (Fig. 1). The $1\text{x}_{\text{g}}\text{-PES}$ is the normal PES; that is, its initial state is the 0_{g} ground state, 1x_{g} . The initial state of the 10_{g}-PES is the 0_{g} first excited state, 10_{g} . All spectra are plotted against the final-state binding energies, which are measured relative to the 0_{g} ground state energy.

The R lines of the $10_{\text{g}}^{-1}1\text{x}_{\text{g}}\text{-AES}$ agree (less than 0.2eV difference) with the bands of the two PES as shown in Fig. 2. This is true not only for the peak positions but also for the line shapes. The broad R1 line (~1.5eV), which Fig. 2 does not show, also agrees with the first band (12.5eV) of the $1\text{x}_{\text{g}}\text{-PES}$ [9]. These agreements imply the following:

- The initial state of the R lines is indeed the $10_{\text{g}}^{-1}1\text{x}_{\text{g}}(^3\text{P}_{\text{g}})$ state.

- The final states of the R lines are all one-hole states, whose electronic configurations, symmetries, and usual designations [9] are listed in Table 1.

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(c) the measured energies, that is, the low-lying $(2p_{1/2})$ state energy (50.8 eV) [1] and the Auger electron kinetic energies measured by Siegbahn et al. [1], are accurate to within 10 eV.

Previously the K lines were assigned by Moddemann et al. [2]. They assigned the K α line to the Fe-W process and the K β -K γ lines to the Fe-W process. However we have found that not only the K α line but also the K β -K γ lines are due to the Fe-W process.

2. Ni-Ni lines

2.1. Various spectra for U_2^{++} states

The U_2^{++} states are reflected in various spectra, such as the Auger Electron Spectrum (AES), the Double Charge Transfer Spectrum (DCTS), and the Electron Impact Mass Spectrum (EIMS). In the following, we shall describe how these three spectra provide U_2^{++} information, emphasizing the selection rules.

2.1.1. Double charge transfer spectrum

When protons of a fixed translational energy collide with U_2 molecules, a single proton can remove two electrons from a molecule and become an H^+ ion. A translational energy distribution of the H^+ ions gives a double charge transfer spectrum.

From the energy conservation for the process,



we have

$$E(U_2^{++}) - E(U_2) = I(H^+) - I(H^-) + E(H^+) - E(H^-) + I(U_2) - I(U_2^{++}) \quad (2)$$

where $I(M)$ denotes the translational energy of M , and $E(M)$, the internal energy of M .

It is shown [11] that the H^- ions are formed in its ground state; therefore, $E(H^-) - E(H^+) = 14.35$ eV. The recoil energy of U_2^{++} is negligible; that is, $I(U_2) = I(U_2^{++})$. Substituting these two equations into Eq. (2), we have

$$E(U_2^{++}) - E(U_2) = I(H^+) - I(H^-) + 14.35 \text{ eV} \quad (3)$$

In this equation, $E(U_2)$ is the U_2 ground state energy, and $I(H^+)$ is a fixed translational energy. Thus, we get the U_2^{++} state energies, $E(U_2^{++})$, by measuring the translational energies of the H^+ ions, $I(H^+)$.

Two single-charge transfers,



the threshold energy, or the ground state energy of O_2^{++} . On the other hand, breaks in the O_2^{++} spectrum give excited state energies of O_2^{++} .

The O_2^{++} states formed via the process (5) can be singlet, triplet, quintet, or septuplet; and can have any symmetries.

The EIMS in Fig. 3 was measured by Daly and Fowell [14]. They used the charge transfer reaction $O_2^{++} + N_2 \rightarrow O_2^+ + N_2^+$ to detect the O_2^{++} ions. The EIMS in Fig. 3 shows the $[O_2^{++}]^{1/2}$ plot against the incident electron energy. The zero of the O_2^{++} yield is indicated by a short horizontal line.

3.1.3. Auger electron spectrum

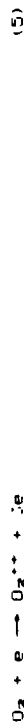
When a core electron, $1s$, or $2s$, is removed from O_2 , the $2s$, $2p$, and $2d$ states are formed. Only the $2p$ and $2d$ states, however, can be populated by removing a core electron from the O_2 ground state $^3\Sigma_g^-$, because of the selection rule based on a single configuration. The binding energies of these two states are 543.1 eV ($2p$) and 544.2 eV ($2d$) [1] and are separated by 1.140 eV [1,15]. When excited with high-energy photons (24 eV) or high-energy electrons (5 keV), these two states have the relative populations, 2.5 ($2p$) : 1 ($2d$) [1,15]. These $2p$ and $2d$ states are the initial states of the K-W Auger processes. To get the O_2^{++} state energy from an Auger transition

also produce the H^- ions. The H^- ion yield by this process, however, depends quadratically on the O_2 gas pressure, while the H^- ion yield by the process (1) depends linearly. From two H^- spectra of different gas pressures, we can learn which peaks are due to the process (1). All DLS peaks in Fig. 1 are due to the process (1) [12].

The O_2^{++} states formed via the process (1) are triplets, because the proton H^+ is a singlet; the O_2 is in its ground state, $^3\Sigma_g^-$; the H^- ion is formed in its ground state, 1S_g ; and the total spin of each side of the process (1) must be the same. According to an approximate symmetry selection rule [12], the probability of forming the $O_2^{++}(^3\Sigma_g^-)$ states via the process (1) is small. Furthermore, there is a trend that the lower the energy of O_2^{++} , the higher the probability of its formation.

3.1.4. Electron impact mass spectrum

In electron impact mass spectroscopy, the O_2^{++} ions produced by the process



are counted as a function of the incident electron energy. Near the threshold energy, the square root of the O_2^{++} yield is a linear function of the incident electron energy [13]. Therefore, an extrapolation of $[O_2^{++}]^{1/2}$ to zero yield gives

energy, we have to know from which initial state the Auger transition starts.

Auger transitions from the $^4\Sigma^-$ and $^2\Sigma^-$ states to an identical final state form a pair of Auger lines that satisfy the following three requirements:

- (a) the separation of the lines is 1.140-eV.
- (b) the widths of the two lines are almost the same.
- (c) the line from the $^4\Sigma^-$ state is much stronger than the line from the $^2\Sigma^-$ state.

Three pairs of Auger lines, (N1,N2), (N5,N6), and (N8,N9), satisfy the above three requirements, as shown in Figure 4. These three Auger-line pairs therefore apparently arise from Auger transitions from the $^4\Sigma^-$ and $^2\Sigma^-$ states to an identical final state.

For an allowed Auger transition, the initial state and the final state that includes the outgoing electron must have the same spin and the same symmetry. The selection rules for the final (ionized) state, which does not include the outgoing electron, are the following:

- (a) the allowed spins of the final states, that decay from the initial state with spin S , are $S-1/2$ and $S+1/2$.
- (b) the allowed symmetries of the final states, that decay from the initial state with symmetry Σ^- (or Σ^+), are any symmetries but Σ^- (or Σ^+).

These selection rules arise because the outgoing electron

orbital, like any other orbital, must have spin $1/2$ and any symmetry but Σ^- . From these selection rules, we can draw two conclusions:

- (a) the final states of the paired Auger transitions must be triplets, because only the transitions, $^4\Sigma^- \rightarrow$ singlet or triplet and $^2\Sigma^- \rightarrow$ triplet or quintet, are allowed.
- (b) the final states of the Auger lines N1-N9 cannot have the Σ^- symmetry, because a $\Sigma^- \rightarrow \Sigma^-$ transition is forbidden.

3.2. Comparison of AES with EIMS and DLIS

Figure 3 compares the $U_{2^{++}}$ state energies obtained from EIMS, DLIS, and AES. The $U_{2^{++}}$ state energies are measured relative to the U_2 ground state energy.

The lowest $U_{2^{++}}$ (35.8eV) state is observed only in EIMS. In DLIS and AES, transitions to this state are forbidden, because this state has the 1E_g ($1g^-$) symmetry [12].

The second lowest $U_{2^{++}}$ (40.8eV) state is observed in EIMS as a peak and in DLIS as a shoulder, but not in AES. In AES, transitions to this state are forbidden, because this state has the 3E_g ($1g^-$) symmetry [12]. In DLIS, we might expect, based on the magnitude of transferred energy, that the first band (40.8eV) should be much stronger than the second band (42.8eV). But in reality, the first

the difference between the two cases is that the first case is a *case of* the second case, while the second case is a *case of* the first case. This is the difference between the two cases.

to obtain the \mathcal{L}_2 stability of the closed-loop system. In this case, a linear feedback K from the y to the u is chosen, which leads to the closed-loop transfer function $T(s)$ from the w to the y given by

As a result of the above, the authors have concluded that the use of the proposed method is not only feasible but also effective in the detection of the faults in the power transformer. The proposed method is simple and easy to implement. The authors are confident that the proposed method will be useful in the detection of the faults in the power transformer.

[illegible]

In the preceding section, we have shown that the observed ^{12}C states are triplets except for the lowest state. The order of the assigned triplet states (table 2) agrees with the order of the lowest and triplet states calculated by Keene et al. (4).

The triplet states that have been observed in the isoelectronic systems N_2 [16,17] and NU^+ [18] correlate well with the triplet states of U_2^{++} , as shown in Figure 5. The U_2^{++} potential curves are constructed from the calculated internuclear distances [3] and the observed state energies. In the N_2 and NU^+ spectra, the states corresponding to $1\pi_u$, $2(\pi_g)$ and $2\sigma_u$, $1(\pi_u)$, $1(\pi_g)$ have not been observed, because these states can be reached only by a two-electron jump from the ground states of N_2 and NU^+ .

Finally the assigned triplet state energies agree very well with those calculated by Hurley [13]. He has not calculated the $U_2^{++}(\pi_g)$, $U_2^{++}(\pi_u)$, and $U_2^{++}(2\sigma_u)$ state energies.

3.3.2. Auger-line widths

An Auger-line width is closely related to the internuclear distance (r_e) of the Auger final state: the smaller the difference between the r_e 's of the Auger final state and the ground state, the narrower the Auger line. The $1\pi_u$, $1(\pi_g)$, $3\sigma_g$, $1(\pi_u)$, and $2\sigma_u$, $1(\pi_u)$ states are expected to have r_e 's similar to the ground state, while the $2\sigma_u$, $1(\pi_g)$ state is expected to have an r_e considerably smaller than the ground state and the $1\pi_u$, 2 state, an r_e much larger than the ground state. (These estimates are based on the $1\pi_u$ and $3\sigma_g$ bonding and $1\pi_g$ and $2\sigma_u$ antibonding characters of the orbitals involved.) Thus, we expect that the N_2 ($1\pi_u$, $1(\pi_g)$), N_2^+ ($3\sigma_g$, $1(\pi_u)$), and N_2^+ ($2\sigma_u$, $1(\pi_u)$)

lines should be narrower than the N_2 ($1\pi_u$, 2) and N_2^+ ($2\sigma_u$, $1(\pi_g)$) lines. This agrees with Figures 1 and 4, supporting our assignments.

Wreghitt and Hayes [6] assigned the $2\sigma_g$, $1(\pi_u)$ configuration to the N_2 line. The $3\sigma_g$, $1(\pi_u)$ configuration is expected to have an r_e considerably larger than the ground state r_e . This implies that the N_2 line should be wider than the N_2^+ line, contrary to Figures 1 and 4. For this reason, we prefer to assign the $2\sigma_u$, $1(\pi_u)$ state to the (N_2 , N_2^+) Auger lines.

Hurley's calculations [3] have shown that the r_e for the $3\sigma_g$, $1(\pi_g)$ state is the closest to that for the ground state; the r_e for the $1\pi_u$, $1(\pi_g)$ states, the second closest; and the r_e for the $2\sigma_u$, $1(\pi_g)$ state, the third closest; as can be seen in Figure 5. This would predict that the N_2 line due to $3\sigma_g$, $1(\pi_g)$ is the narrowest; the N_2 and N_2^+ lines due to $1\pi_u$, $1(\pi_g)$, the second narrowest; and the N_2 line due to $2\sigma_u$, $1(\pi_g)$ state, the third narrowest among these lines. This ordering of the line widths agrees with the line widths observed in Figure 4.

3.3.3. Auger line intensities

Figure 4 shows that the N_2 and N_2^+ intensity due to $1\pi_u$, $1(\pi_g)$, the N_2 intensity due to $1\pi_u$, 2 , and the N_2 and N_2^+ intensity due to $2\sigma_u$, $1(\pi_u)$ are much stronger than the N_2 intensity due to $3\sigma_g$, $1(\pi_g)$ or the N_2 intensity due to $2\sigma_u$,

NB and H9 lines, (Subsection 3.3.2), and smaller Anger line intensities for the strong NB and H γ lines (Subsection 3.3.3), all inconsistent with observations.

Diuldas et al. [5] assigned the paired Auger lines

Gregoritch and Hayes (6) assigned the N2 line to $1\pi_g^{-1}3\pi_u^{-1}$ and the N4 line to $3\sigma_g^{-1}1\pi_u^{-1}$. Like the Dunlap's calculations, their χ_e calculations cannot distinguish the individual multiplet states. We agree with the first assignment, but disagree with the second assignment for the reasons given in Subsection 3.3.2.

A summary of our assignments for the Ni-Ny Auger lines are listed in Table I.

The S1 and S2 lines, unlike the R1-R7 lines, appear in the photon-excited ($h\nu=1487\text{eV}$) AES [2]. This implies that

The S1 and S2 lines, unlike the R1-R7 lines, appear in the photon-excited ($h\nu=1487\text{eV}$) AES [2]. This implies that

the B1 and B2 lines do not arise from a 1e-W or 1e-WW euger process, but arise from a 1e-W or 1e-WW euger process.

4.1. Previous assignments of the B1 and B2 lines

Moddemann et al. [1], Siegbahn et al. [11], and Gregoritch and Hayes [6] have assigned the B1 and B2 lines to the euger transitions from the $2L(10^+)$ and $4E(10^+)$ states to the $1B_0^+(1X_0^+)$ state. This assignment is not acceptable for the following reasons:

- the separation (1.8eV) of the B1 and B2 lines does not agree with the separation (1.1eV) of the initial states.
 - the relative intensity (B2:B1) of the B1 and B2 lines is not like that of the other paired euger lines (B1:B2 = 0.7:0.3).
 - the $1B_0^+(1X_0^+)$ state energy (8.1eV) or B2 line as determined by their B1 or B2 assignment respectively does not agree with that (5.8eV) measured by electron impact mass spectrometry [14,15,16,17]. We believe that the vertical $1B_0^+(1X_0^+)$ state energy is (6.2eV).
 - the euger transition from the $2L^+$ initial state to the $2L^+$ final state is forbidden (see subsection 4.1.3).
- Realizing that euger transitions from quartet to singlet states are forbidden, Bonlap et al. [5] reassigned

the B1 line to the euger transition from the $1B_0^+(10^+)$ state to the $1B_0^+(1X_0^+)$ state and the B2 line to a 1e-W or a 1e-WW euger process. This assignment is also not acceptable for the following reasons:

- the resulting $1B_0^+(1X_0^+)$ state energy (9.9eV) is far too large in comparison with the experimental energy (6.1eV).
- the euger transition from $2L^+$ to $1B_0^+$ is forbidden as well as the euger transition from $4L^+$ to $1B_0^+$ (see subsection 4.1.3).
- the B1 line does not arise from a 1e-W or a 1e-WW euger process, because it appears in the photoionized (single-electron) PES.

4.2. Our assignment of the B1 and B2 lines

We suspect that the B1 and B2 lines are not due to a 1e-W process but due to a 1e-WW process. The core level PES (level of 0e) shows pronounced satellite (shake-up) peaks at 15.0eV and 16.1eV above the main $2L(10^+)$ peak. We attribute the B1 and B2 lines to the euger transitions from these two shake-up states to the $1B_0^+(1X_0^+)$ state. The reasons for this B1/B2 assignment are as follows:

- the separation (1.8eV) of the shake-up peaks agrees with the separation (1.8eV) of the B1 and B2 lines.
- the relative intensity (B1) of the shake-up peaks agrees with the relative intensity (B2) of the

S1 and S2 lines.

- (c) The relative magnitude between the S1 and the N1-N9 lines is comparable to the relative magnitude between the shake-up and the main peaks.
- (d) Among the Auger decays starting from the shake-up state ($10^{-1}1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1}$), the decay into the $U_{2^{++}}(1\alpha_{\frac{1}{2}}^{-1}2^{-})$ state is expected to be the strongest. (This will be shown in the following paragraph.)
- (e) The $U_{2^{++}}(1\alpha_{\frac{1}{2}}^{-1}2^{-})A_0$ state energy (47.2eV) calculated from the (N5,N6) lines agrees with that (47.7eV) calculated from the (S1,S2) lines. The small difference (0.4eV) between the two calculations is due to the different intermediate states, the main (10^{-1}) in the former and the shake-up ($10^{-1}1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1}$) state in the latter.
- (f) A 14eV-WV Auger transition has been observed in the AES of UO (23).

The 2h final states that can be reached from the initial state $10^{-1}1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1}$ by an allowed Auger transition are $1\alpha_{\frac{1}{2}}^{-1}V^{-1}$, where V can be any of $1\alpha_{\frac{1}{2}}^{-1}$, $1\alpha_{\frac{1}{2}}^{-1}2\alpha_{\frac{1}{2}}^{-1}$, or $2\alpha_{\frac{1}{2}}^{-1}$. The corresponding Auger transitions are:

$$(1\alpha_{\frac{1}{2}}^{-1}V)^{-1} \rightarrow (10,0) \quad \text{for } V = 1\alpha_{\frac{1}{2}}^{-1}, 1\alpha_{\frac{1}{2}}^{-1}2\alpha_{\frac{1}{2}}^{-1}, 2\alpha_{\frac{1}{2}}^{-1} \quad (6)$$

where 0 stands for the outgoing electron. The σ^{-1} -AES shows that, among the Auger transitions (6), the transition with $V=1\alpha_{\frac{1}{2}}^{-1}$ is by far the strongest. (See table 2 and Fig. 4, and

recall that the other strong Auger transitions are $(1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1})$ and $(1\alpha_{\frac{1}{2}}^{-1}2\alpha_{\frac{1}{2}}^{-1})$.) Therefore, among the Auger decays starting from the shake-up ($10^{-1}1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1}$) state, the decay into the $1\alpha_{\frac{1}{2}}^{-1}2^{-}$ state is expected to be the strongest. In fact, according to our interpretation, we have observed this decay but not other decays.

Our assignments of the S1 and S2 Auger lines are included in table 1. Note that the order of the $4\Sigma^{-}$ and $2\Sigma^{-}$ states arising from $10^{-1}1\alpha_{\frac{1}{2}}^{-1}1\alpha_{\frac{1}{2}}^{-1}$ is reversed from that of the $4\Sigma^{-}$ and $2\Sigma^{-}$ states arising from 10^{-1} . The reason is given in the Appendix.

5. NS1 - NS3 lines

The broadness of the NS1-NS3 lines is due to CI mixing of the 2h states with 3h-1e states. A large number of 3h-1e states are expected in this energy region. The majority of them are Rydberg states converging to 2h states and are densely distributed. Auger transitions from the $0^{-1}(10^{-1})$ states into these 3h-1e states are forbidden, since they are three-electron-jump transitions. The intensities of the NS1-NS3 lines are due to 2h states, and the peaks of the NS1-NS3 lines tend to match with the 2h state energies prior to the CI mixing. In this section, we assign the pre-CI 2h states to the NS1-NS3 lines.

Figure 6 compares the $U_{2^{++}}$ -AES (1) with the two NO-AES (2), the nitrogen core ($1s_{N^{-1}}$) and the oxygen core ($1s_{O^{-1}}$)

spectra. A sum of the two NU-AES almost reproduces the O_2^- AES, as can be seen in Figure 6. The three NS1-NS3 peaks are clearly reproduced by this addition. Our assignment for the NS1-NS3 lines are based on this correspondence between the NU-AES and the O_2^- -AES.

For the NU-AES assignments, we can use the intensity changes between the $1s_{2,1}$ and $1s_{1,1}$ spectra. Furthermore, there is an accurate calculation [24] by the CI method for NU-AES. Our assignments of the NU-AES, details of which will be published elsewhere, are shown in the bottom of Figure 6.

Based on the close correspondence between the NU-AES and the O_2^- -AES, we assign $2a_u^{-1}3a_u^{-1}$ and $2a_g^{-1}1x_u^{-1}$ to the NS1 line, $2a_g^{-1}3a_g^{-1}$ to NS2, and $2a_g^{-1}2a_u^{-1}$ to NS3. Since the assignments are tentative, they are not included in Table 1.

Previously Dunlap et al. [5] assigned $2a_u^{-1}3a_g^{-1}$ to NS1, $2a_g^{-1}1x_u^{-1}$ to NS2, and $2a_g^{-1}2a_u^{-1}$ to NS3. Also recently Gregoritch and Hayes assigned $2a_u^{-1}2$ to NS1, $2a_g^{-1}1x_u^{-1}$ to NS2, and $2a_g^{-1}2a_u^{-1}$ to NS3. The three studies agree on the NS3 assignment, but disagree on the NS1 and NS2 assignments. We believe that the energy of $2a_u^{-1}3a_g^{-1}$ is too low for the NS1 assignment. Also, the $2a_g^{-1}1x_u^{-1}$ states may not be observed as a resolved peak, because eight allowed states arise from the $2a_g^{-1}1x_u^{-1}$ configuration.

6. Summary

Our assignments of the O_2^- Auger electron spectrum are summarized in Table 1. These assignments are confirmed by selection rules, calculations, and other experiments. The assignments of the NS1-NS3 lines are tentative; they are not included in Table 1.

Appendix: Shake-up states of $O_2^+(1g_{g,1})$

Since shake-up states have three configurations, photoionization into shake-up states are forbidden. Shake-up states borrow their photoionization intensity from the $1h$ states, $2E^-(1g_{g,1})$ and $4E^-(1g_{g,1})$, through CI mixing. The observed shake-up states in FES, therefore, must have either the $2E^-$ or $4E^-$ symmetry.

The lower-lying shake-up states are expected to arise from the electronic configurations, $1g_{g,1}^{-1}1x_g$, $1g_{g,1}^{-1}2a_g^{-1}1x_g$, or $1g_{g,1}^{-1}2a_u^{-1}1x_g$. The first configuration gives both the $2E^-$ and $4E^-$ symmetries, but the second and the third configurations do not give either symmetry. The lowest observed shake-up state, therefore, must arise from the $1g_{g,1}^{-1}1x_g$ configuration.

The $1g_{g,1}^{-1}1x_g$ configuration gives two $2E^-$ states and one $4E^-$ state. In Fig. 7, the relative positions of these three states are estimated from the observed $O_2(1x_u, 1x_g)$ states, c $1E_u$ and B $1E_u$ (25,26). Coupling with the core-hole $1g_{g,1}$, the $2E^-$ states (B $1E_u$) and the ground state $1E_g$

split into the $^4\Sigma^-$ and the $^2\Sigma^-$ states. Assuming that the two splittings are the same and that the weighted average of the $^4\Sigma^-$ (4^+) and $^2\Sigma^-$ (2^+) state energies gives the $^2\Sigma^-$ energy, we have obtained the relative positions of the $^4\Sigma^-$ and $^2\Sigma^-$ states. The results are shown in Figure 2.

Two states tend to mix more in Cl, when they are closer in energy. Based on this trend, we expect that the lower $^4\Sigma^-$ state is stronger in intensity than the higher $^4\Sigma^-$ state. The $^4\Sigma^-$ state mixes less than the lower $^2\Sigma^-$ state, but the $^4\Sigma^-$ state has the higher multiplicity than the lower $^2\Sigma^-$ state. Considering these factors, we expect that the lower $^2\Sigma^-$ state and the $^4\Sigma^-$ state have about the same intensities in PES and that the higher $^4\Sigma^-$ state is much weaker than the other two.

To sum up, two shake-up peaks due to the lower $^2\Sigma^-(10^-11\lambda_0-11\lambda_0)$ and the $^4\Sigma^-(10^-11\lambda_0-11\lambda_0)$ states are expected around 6.9eV and 8.6eV above the main $^4\Sigma^-(10^-11)$ peak. These peaks are expected to have similar intensities and to be well separated from the next lowest shake-up peak.

Shake-up peaks are observed at 8.9eV and 10.7eV above the main $^4\Sigma^-(0^-1)$ peak [1,22]. These peaks have similar intensities and are well separated from the next lowest shake-up peak, which appears around 22eV above the main peak [1,22].

Our estimates and the experiments agree well, especially on the separation of the two peaks: 1.7eV (estimated) and 1.8eV (observed). Based on this agreement, we assign the lower $^2\Sigma^-(10^-11\lambda_0-11\lambda_0)$ state to the 8.9eV peak and the $^4\Sigma^-(10^-11\lambda_0-11\lambda_0)$ state to the 10.7eV peak.

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Table 1
Assignment of O₂ Auger electron spectrum

Band	h _ν , E ₂ (eV)	Initial state Config.	Initial state Design.	Final state Config.	Final state Design.	B ₁ E ₁ (eV)
R1	518.	10 _g ⁻¹ 1x _g	2Π _g	1x _g ⁻¹	X 2Π _g	13.
R2	514.1	"	"	1x _g ⁻¹	a 2Π _g	16.7
R3	513.1	"	"	"	A 2Π _g	17.7
R4	512.2(4)	"	"	20 _g ⁻¹	b 2Π _g	18.6
R5	511.7	"	"	1x _g ⁻¹	20 _g	19.1
R6	510.9(4)	"	"	20 _g ⁻¹	2Δ _g	19.9
R7	510.4(4)	"	"	"	B 2Π _g	20.4
S1	506.1(5)	10 ⁻¹ 1x _g ⁻¹ 1x _g	2Π _g	1x _g ⁻²	2Δ _g	47.7
S2	504.3(4)	"	2Π _g	"	"	"
N1	501.6(3)	10 ⁻¹	2Π _g	1x _g ⁻¹ 1x _g ⁻¹	W 2Δ _g	42.6
N2	500.5(3)	"	2Π _g	"	"	"
N3	499.7(3)	"	"	20 _g ⁻¹ 1x _g ⁻¹	b 2Π _g	43.4
N4	499.3(4)	"	"	1x _g ⁻¹ 1x _g ⁻¹	b' 2Π _g	43.8
N5	496.9	"	2Π _g	1x _g ⁻²	2Δ _g	47.3
N6	495.8(4)	"	2Π _g	"	"	"
N7	494.6(4)	"	"	20 _g ⁻¹ 1x _g ⁻¹	c 2Π _g	48.5
N8	492.9(4)	"	2Π _g	20 _g ⁻¹ 1x _g ⁻¹	2Π _g	51.2
N9	491.9(4)	"	2Π _g	"	"	"

Table 2
Observed O₂⁺⁺ states

Binding Energy (eV)	Electronic Configuration	Designation & Symmetry	Spectroscopy AES	DCIS	EIMS
36.3(5)	1x _g ⁻²	X 1Σ _g ⁺	f _g	f	threshold
40.8(2)	1x _g ⁻¹ 1x _g ⁻¹	a 2Σ _g ⁺	f	shoulder	break
42.6	1x _g ⁻¹ 1x _g ⁻¹	W 2Δ _g	N1, N2	peak	break (?)
43.4	20 _g ⁻¹ 1x _g ⁻¹	b 2Π _g	N3	shoulder	---
43.8	1x _g ⁻¹ 1x _g ⁻¹	b' 2Σ _g ⁺	N4	shoulder	---
47.3	1x _g ⁻²	2Δ _g	N5, N6	peak	---
48.5	20 _g ⁻¹ 1x _g ⁻¹	c 2Π _g	N7	shoulder	---
51.2	20 _g ⁻¹ 1x _g ⁻¹	2Π _g	N8, N9	shoulder	---

a) forbidden transition

Figure captions

- Fig. 1. Electron-impact Auger electron spectrum for O_2 obtained by Siegbahn et al. [1]. The K α band is taken from Ref. [2]. The labelled lines are assigned in Table 1.
- Fig. 2. Comparison of the low ^{14}N -AES [1] with the ^{14}N -AES [9] and the ^{14}N -AES [10]. All three spectra are plotted against the O_2 final-state energies, which are measured relative to the O_2 ground state energy.
- Fig. 3. Comparison of the ^{14}N -AES [1] with the Electron Impact Mass Spectrum [14] and the Double Charge Transfer Spectrum [12]. All three spectra are plotted against the O_2 state energies, which are measured relative to the O_2 ground state energy. The EIMS shows the square-root plot of the O_2 ion yield. The O_2 state energies of AES are based on the initial state assignments given in Table 1.
- Fig. 4. Paired Auger transitions from the ^{14}N - and ^{14}N -initial states to an identical final state. The paired Auger transitions are connected at the bottom. The shaded area indicates the N_2 - N_2 portion of the spectrum. This spectrum is an expansion of Figure 1.

- Fig. 5. Comparison among the isoelectronic states of N_2 , NO^+ , and O_2^{2+} . The broken-line potential curve is the $b^1\Pi_g$ state. The state names used in the NO^+ system [18] are adopted here. The potential curves of N_2 (16,17) and NO^+ (18) are from experiments. The O_2^{2+} potential curves are constructed from the calculated internuclear distances [3] and the observed state energies. The vertical broken line for O_2^{2+} shows the internuclear distance of the O_2 ground state. The vertical broken lines for the N_2 and NO^+ systems are drawn at the same relative positions as the O_2^{2+} system for comparison.
- Fig. 6. Comparison of the O_2 -AES [1] with the two NO -AES [2], the nitrogen core (15 π^*) and the oxygen core (15 σ^*) spectra. Our assignments of the NO -AES are given in the bottom.
- Fig. 7. Estimation of the ^{14}N -AES [14] and ^{14}N -AES [14,15] state energies from the observed ^{14}N -AES [14,15] and ^{14}N -AES [14,15] state energies. The following experimental data are used in the diagram: the vertical excitation energy (8.6eV) of the ^{14}N state [15]; the energy difference (1.1eV) of the ^{14}N and the ^{14}N states [15]; the energy difference (1.1eV) of the main ^{14}N (a) and ^{14}N (b) peaks [1,15].

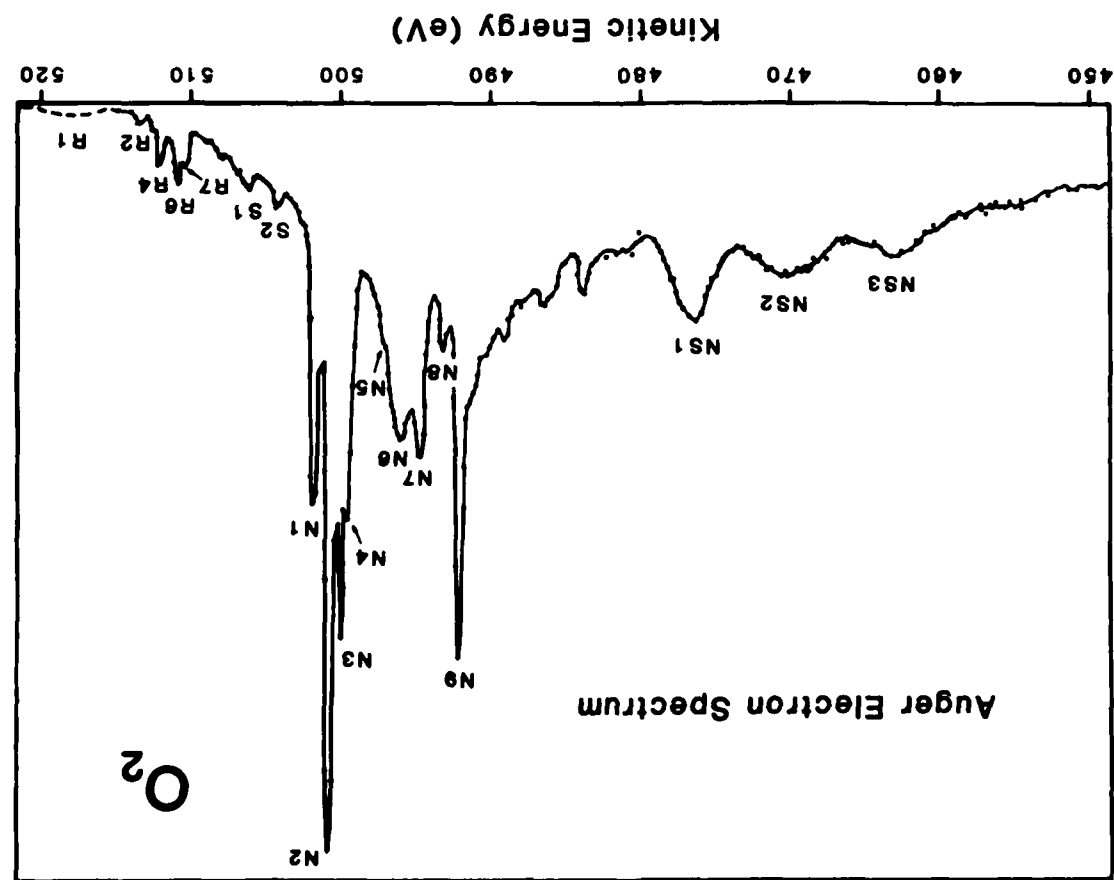


Fig. 1

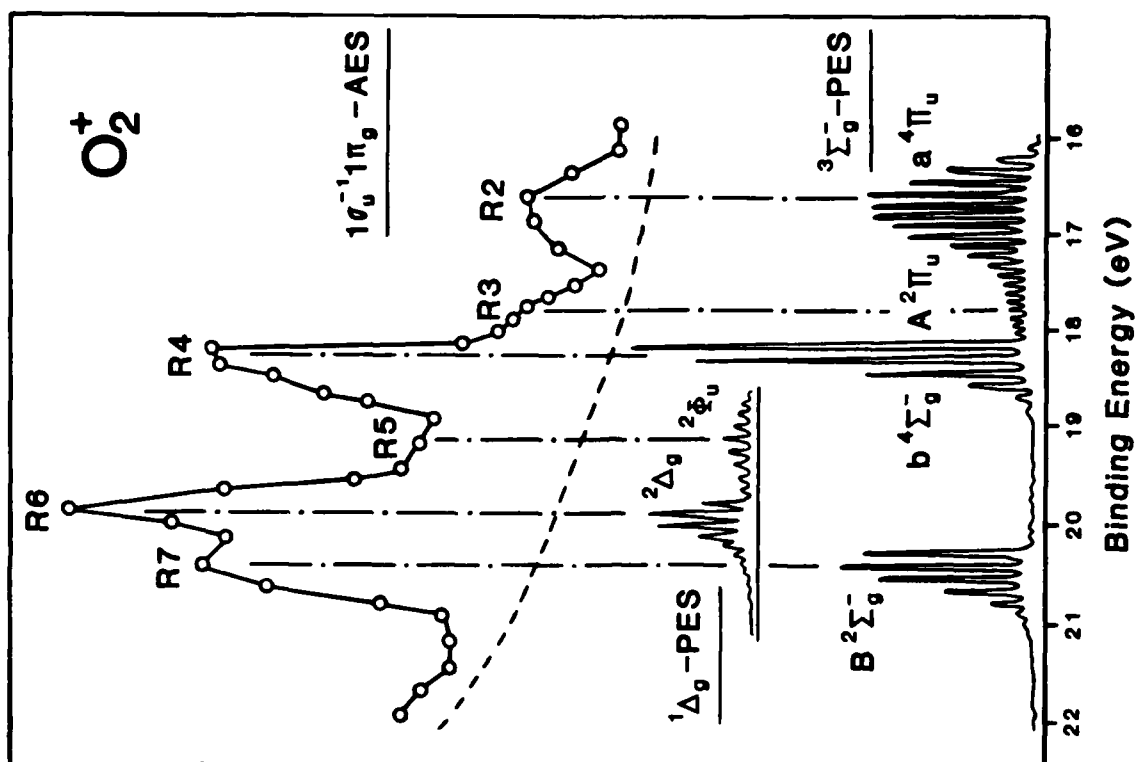
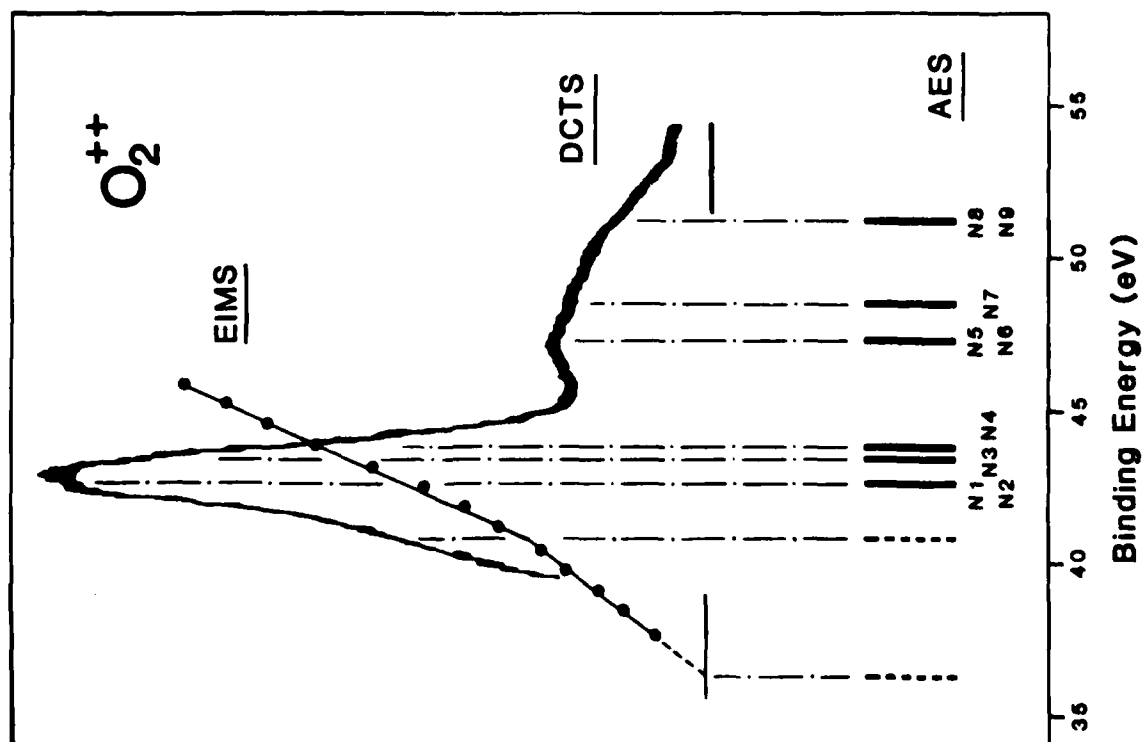
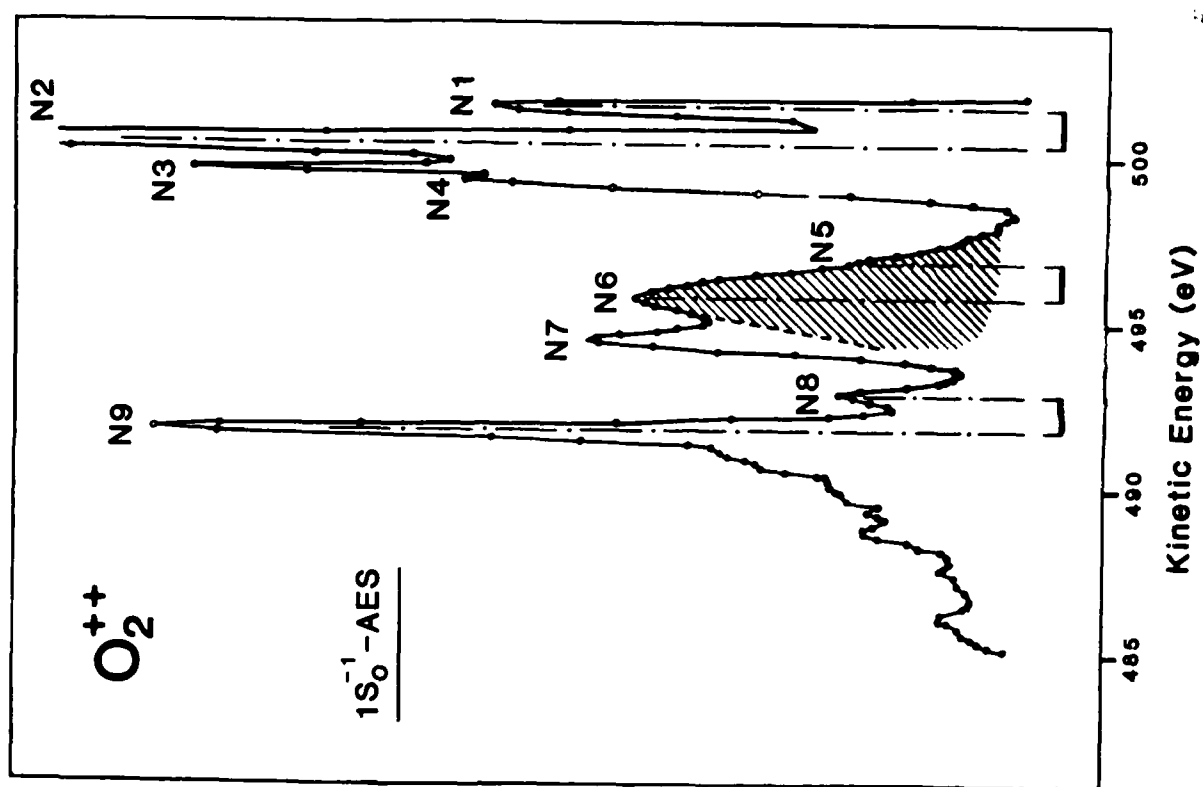


Fig. 2



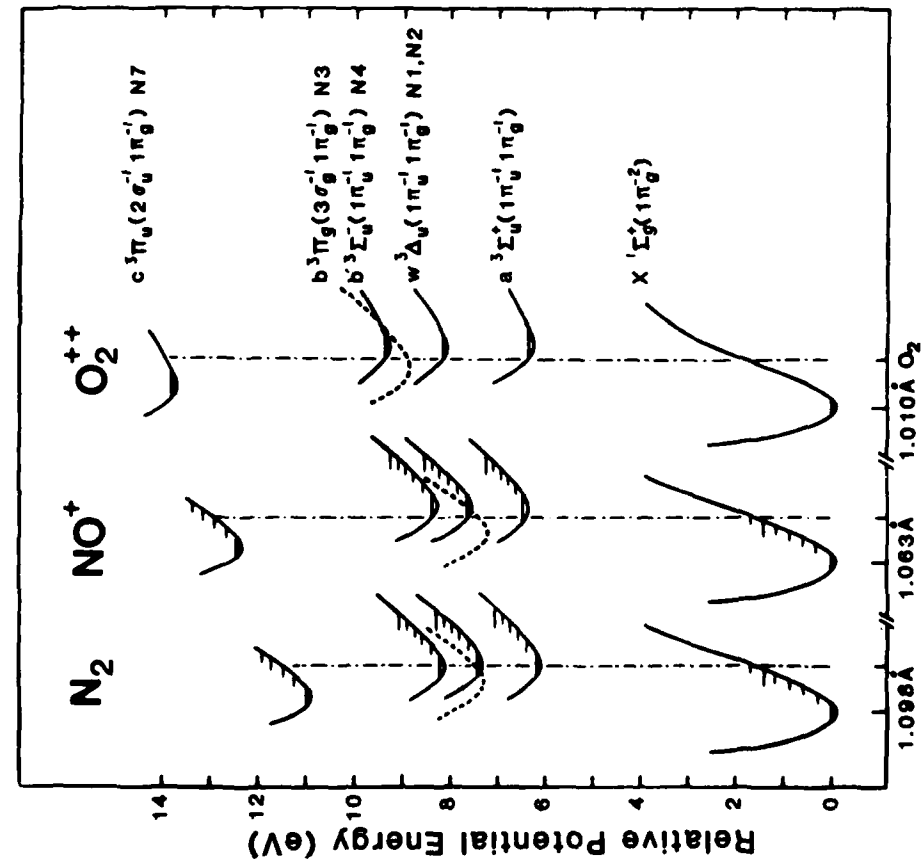


Fig. 5

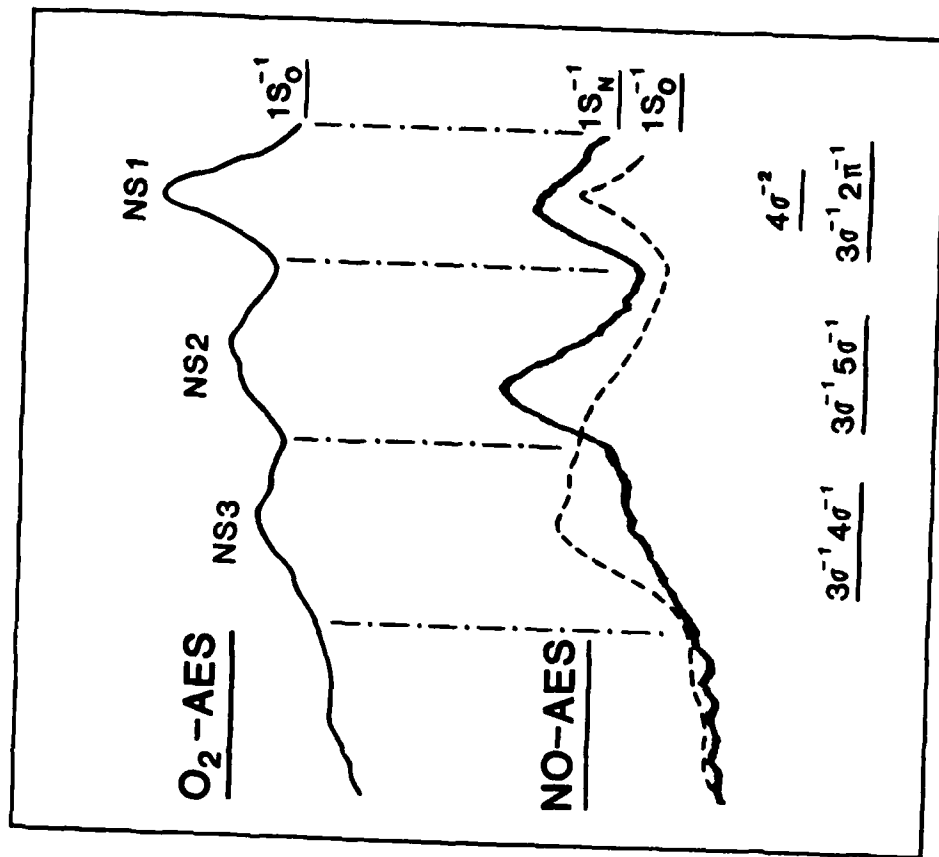
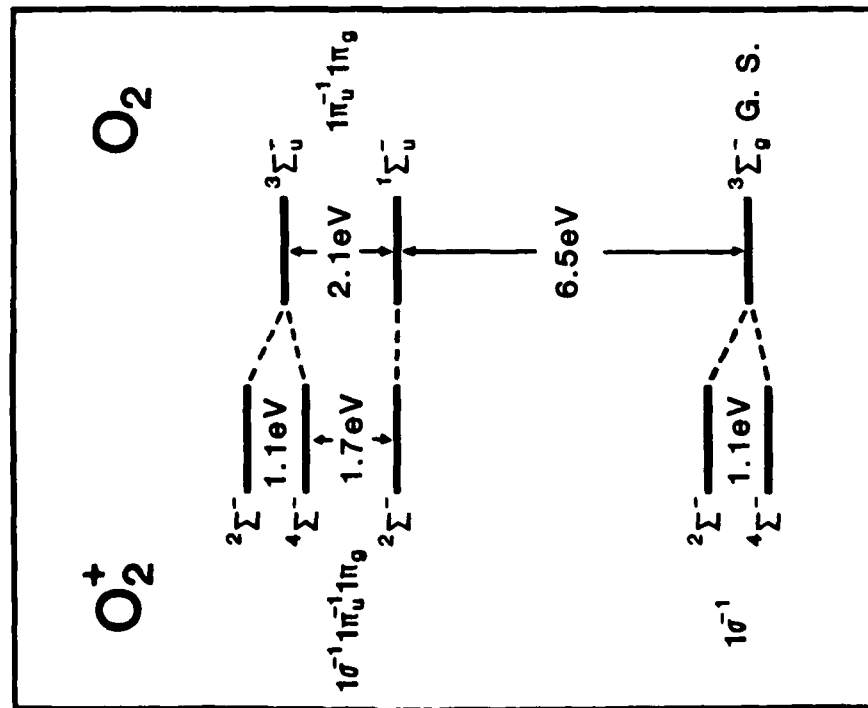


Fig. 6



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